

Epi-6-aminopenicillanic Acid and Epipenicillin G

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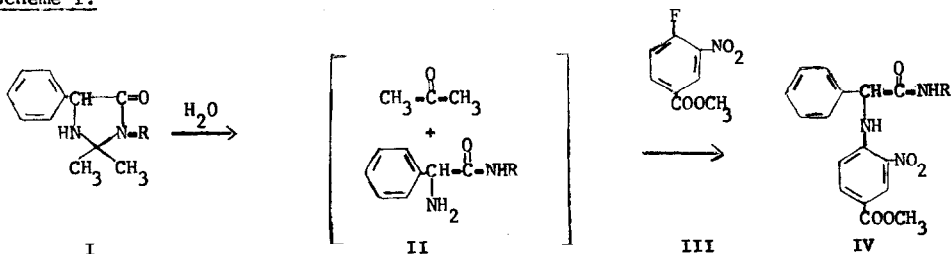
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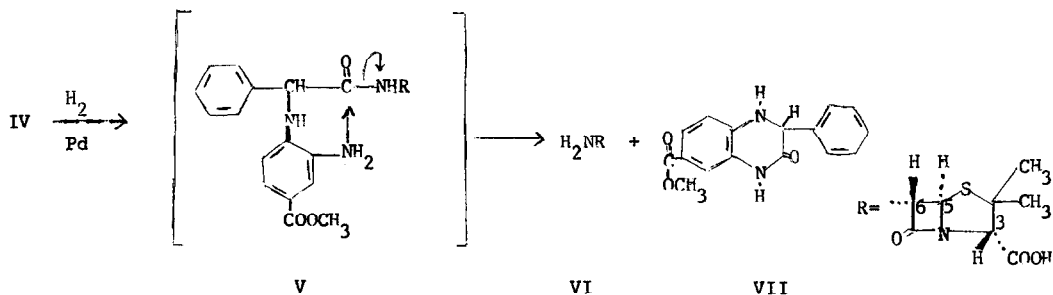
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Recently we reported the first successful epimerization of an intact penicillin nucleus.<sup>1</sup> Our work involved the epimerization of hetacillin and methyl phthalimidopenicillanate.<sup>1,2</sup> Nmr studies and deuterium exchange experiments showed that the C-6 position of the penicillin nucleus had epimerized. The *cis* → *trans* conversion of neighboring protons in β-lactams is accompanied by a numerical decrease of the coupling constant. For monocyclic β-lactams, J (*cis*) varies between 4.9 and 5.9 cps, J (*trans*) between 2.2 and 2.8 cps<sup>3,4</sup>; for bicyclic derivatives (penicillin and cephalosporin) the corresponding values are J (*cis*)=4-5 cps<sup>5</sup> and J (*trans*)=1.5-2 cps.<sup>6-10</sup>

We now wish to report the preparation of epi-6-APA and epibenzylpenicillin (epipenicillin G). After a few fruitless exploratory attempts to epimerize 6-APA and penicillin G, we turned our attention to a less direct but more successful route starting with the readily available epihetacillin.<sup>1</sup> The removal of the side chain was accomplished via a reaction sequence shown in Scheme I. This method was described by R. W. Holley and A. D. Holley<sup>11</sup> for the stepwise removal of amino acids from peptides. Johnson et. al.<sup>12</sup> adopted a slightly modified scheme for generating 6-APA from certain penicillin derivatives.

Scheme I:





Alkylation of epihetacillin with methyl 4-fluoro-3-nitrobenzoate<sup>13</sup> (III) in basic aqueous tetrahydrofuran gave crude (IV) in 83% yield. Epiampicillin (II) appears to be an intermediate in this reaction. Hydrogenation of (IV) in neutral aqueous solution using 30% palladium on alumina as catalyst presumably yielded the amine (V) but no attempt was made to isolate it. Spontaneous cyclization of (V) in a moist, weakly acidic methyl isobutyl ketone extract at room temperature gave crystalline epi-6-APA in 15% yield; m.p. 168-170°,  $[\alpha]_D^{25} + 262.4^\circ$  (c 0.5, water). Anal. Calcd for  $C_8H_{12}N_2O_3S$ : C, 44.43; H, 5.59; N, 12.95. Found: C, 44.46; H, 5.80; N, 13.14.  $\nu$  max. (KBr) 3600, 3400, 1785 ( $\beta$ -lactam CO) and 1620 + 1410  $cm^{-1}$  ( $COO^-$ ). The nmr spectrum of epi-6-APA (60 Mc in  $D_2O$ ) shows singlets at  $\delta$  1.57 and  $\delta$  1.50 ppm (gem dimethyl), further a singlet at  $\delta$  4.39 ppm (H=3), a doublet at  $\delta$  4.67 ppm with  $J=1.6$  cps (H=6) and a doublet at  $\delta$  5.49 ppm with  $J=1.6$  cps (H=5). The small coupling constant is in agreement with a trans  $\beta$ -lactam.

The byproduct 7-carbomethoxy-3,4-dihydro-3-phenylquinoxaline-2-one (VII) was isolated from the mother liquor in 13% yield; m.p. 215°,  $[\alpha]_D^{25} + 34.3^\circ$  (c 1, methanol). Anal. Calcd for  $C_{16}H_{14}N_2O_3$ : C, 68.07; H, 4.99; N, 9.92. Found: C, 67.96; H, 5.19; N, 10.24.  $\lambda$  max. (in methanol) 219  $m\mu$  ( $\log \epsilon$ : 3.836), 241  $m\mu$  ( $\log \epsilon$ : 3.829) and 314  $m\mu$  ( $\log \epsilon$ : 3.654).  $\nu$  max. (KBr) 3265, 1705 (unsat. ester) and 1687  $cm^{-1}$  (amide). The dihydroquinoxaline derivative is identical with a sample prepared by the coupling of D(-)-phenylethylglycine with methyl 4-fluoro-3-nitrobenzoate followed by reduction and ring closure.

In solution, compound VII oxidizes easily to the corresponding quinoxalone derivative:<sup>14</sup> m.p. 280°, Anal. Calcd for  $C_{16}H_{12}N_2O_3$ : C, 68.57; H, 4.31; N, 9.99. Found: C, 68.58; H, 4.33; N, 10.35. The physical data are in agreement with the structure.

Acetylation of epi-6-APA with phenylacetyl chloride yielded epipenicillin G, isolated as the crystalline potassium salt in 71% yield; m.p. 153-154°,  $[\alpha]_D^{25} + 196.4^\circ$  (c 1, water).

Anal. Calcd for  $C_{16}H_{17}N_2O_3 \cdot SK \cdot 1.5 H_2O$ : C, 48.29; H, 5.01; N, 7.02. Found: C, 48.29; H, 5.25; N, 7.35.  $\nu$  max. (KBr): 3600-3100 (hydrate OH), 3320 (NH), 2980 + 2920 ( $CH_3$ ), 1760 ( $\beta$ -lactam CO), 1668 + 1553 (amide), 1620 + 1540 (C=C,  $COO^-$ ) and 702  $cm^{-1}$  (phenyl). The nmr spectrum of the potassium benzylepipenicillinate (60 Mc in  $D_2O$ ) shows a singlet at  $\delta$  1.50 ppm and  $\delta$  1.57 ppm (gem dimethyl), a singlet at  $\delta$  3.62 ppm ( $=CH_2=$ ), further a singlet at  $\delta$  4.32 ppm (H-3), a doublet at  $\delta$  4.80 ppm with  $J=1.6$  cps (H-6), a doublet at  $\delta$  5.25 ppm with  $J=1.6$  cps (H-5) and a singlet at  $\delta$  7.32 ppm (phenyl). The low coupling constant for H-5 and H-6 indicates a trans  $\beta$ -lactam. Potassium benzylepipenicillinate (potassium epipenicillin G) exhibits negligible antimicrobiological activity against standard test organisms.

Esterification of benzylepipenicillin with diazomethane gave crystalline methyl benzylepipenicillinate in 83% yield; m.p.  $116^\circ$ ,  $[\alpha]_D^{25} + 119.1^\circ$  (c 1, chloroform). Anal. Calcd for  $C_{17}H_{20}N_2O_4S$ : C, 58.59; H, 5.78; N, 8.04. Found: C, 58.55; H, 5.99; N, 8.13. The infrared spectrum is similar to that of methyl benzylpenicillinate. The  $\beta$ -lactam CO band is shifted slightly upward to 1785  $cm^{-1}$ . The nmr spectrum (60 Mc, deuteriochloroform) shows a characteristic double doublet at  $\delta$  5.01 ppm with  $J=1.8$  cps (H-6) and a doublet at  $\delta$  5.13 ppm with  $J=1.8$  cps (H-5).

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